



#### Introduction

The HeatLink PEX products are constructed with a unique high density polyethylene and uses special additives to prevent thermal degradation at high temperatures and inhibit UV degradation. Tubing used for radiant floor heating systems may be coated with an optional oxygen diffusion barrier to assist with inhibiting corrosion. The HeatLink production facility has invested in state-of-the-art quality control systems that continuously monitor product quality in real time in addition to the strict quality control testing performed in the two in-house laboratories. Independent testing and certification agencies assure the high quality of the product and conformity to standard requirements as outlined by ASTM International, NSF International, and CSA International to name a few.

HeatLink cross-linked polyethylene tubing exhibit important qualities:

- Minimum bending radii of 6× the tubing diameter at 68°F (20°C)
- Unique flexibility
- Unique chemical resistance
- UV resistance
- · Low friction coefficient
- · Excellent abrasion resistance
- Very good resistance to heat
- Long useful life
- · Low creep rates

## The Material

Polyethylene was discovered in Britain in the 1930's by a series of mistakes and observations made at the ICI Laboratories who were investigating new high pressure reaction types. The raw material, ethylene, is a gas compound that is prepared from petroleum. Under high pressure, and in the presence of oxygen and a catalyst, a polymerization reaction takes place and the small ethylene molecules are linked together creating large chain molecules. These chain molecules are very long and can contain hundreds or thousands of ethylene units. The material resulting from this chemical process is our base plastic material, polyethylene.

Depending on the polymerization process, polyethylene is divided into three groups, low-density PE (LDPE), medium-density PE (MDPE) and high-density PE (HDPE). Due to their denser crystalline structure, HDPE and MDPE show better mechanical and thermal properties than LDPE.

Non-cross-linked polyethylene and polybutylene belong to a group of plastic materials called thermoplastics. The term thermoplastic describes the polymer's ability to be softened at high temperature and then formed by extrusion or injection molding. When the material is cooled, the form given to it during the molding process is retained. However, if the material is heated and cooled repeatedly, it begins to

lose its mechanical properties until the product can no longer fulfill its purpose. Thermosets, as opposed to thermoplastics, are materials that cannot be softened by heat once they are formed. These materials include the unsaturated polyester resins, epoxy, phenolic resins, and crosslinked polyethylene.

The most desirable product between thermosets and thermoplasts would be one that contains the characteristics and advantages of both types of plastics. This would create a thermoplast that could be formed by a simple molding process like extrusion, and then could be crosslinked to produce the characteristics and advantages of the thermoset plastics. This is exactly what is achieved by crosslinking polyethylene. During cross-linking, the individual polyethylene molecules are linked together, creating a network which improves on the properties of ordinary, uncrosslinked polyethylene tubing and giving it the characteristics and advantages of both thermosets and thermoplastics. Crosslinking causes the retention of the desired PE properties at higher temperatures and enhancement of the room temperature properties. After crosslinking the polyethylene changes from a thermoplastic to a thermoset plastic. This plastic has a very stable molecular structure, enabling it to withstand adverse temperature and pressure conditions and has an unsurpassed resistance to abrasion and chemical attack.



# **PEX Tubing Standards & Testing**

Most materials tend to lose part of their physical and mechanical properties over time. This loss is caused by changes to the chemical structure as the product begins to break down. The chemical change can be caused by many different factors such as UV radiation, elevated temperatures, or mechanical stress. The problem is accelerated if the material is exposed to an aggressive environment for extended periods of time, which can lead to a decline in some physical properties such as tensile strength, flexible strength and impact strength.

Within the big family of plastic materials the thermoplasts are the most sensitive to aging and show, with the passage of time, a sharp decline in strength that causing a quicker failure of the material. Crosslinked polyethylene, which is thermoset material, is characterized by extremely high long-term stability. In the graph below, long-term pressure and temperature tests show that PEX tubing has a very gradual weakening of the tubing wall over an extended period of time. This is in contrast to polybutylene which shows a more dramatic loss of performance over time. A sharp change in direction in the graph (circled) indicates that a weakening of the tubing wall has taken place, as in the case of polybutylene.

To ensure the long-term stability of the HeatLink® PEX products, the tubing is tested and certified to many different industry accepted performance standards from agencies such as ASTM International (www.astm.org), CSA International (www.csa-international.org), NSF International (www.nsf.org), International Association of Plumbing and Mechanical Officials (IAPMO) (www.iapmo.org), and the Plastics Pipe Institute (www.plasticpipe.org).

# **HeatLink® PEX-a Tubing**

# Standards ANSI/NSF-14 ANSI/UL 263 ASTM F876 ASTM F877 ASTM F1807 ASTM F1960 ASTM F2080

ASTM F2098 ASTM F2159 ASTM F2657 ASTM E84

CAN/ULC S101 (all but %") CAN/ULC S102.2 CSA B137.5

DIN 4726/9 (EVOH oxygen barrier)

# Listings

cNSF®us-rfh – PEX 0306 ICC-ES PMG 1087 IAPMO UMC PPI TR-4 CSA B137.5 RFH cQAlus P371 UL/ULC

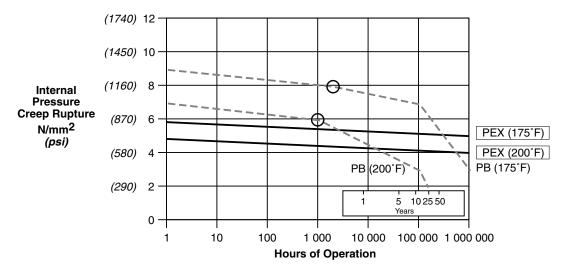
# PureLink® Plus PEX-a Tubing

# Standards

ANSI/NSF-61 ANSI/NSF-14 ANSI/NSF-372 ANSI/UL 263 ASTM F876 ASTM F877 **ASTM F1807 ASTM F1960** ASTM F2023 **ASTM F2080 ASTM F2098 ASTM F2159 ASTM F2657 ASTM E84** CAN/ULC S101 CAN/ULC S102.2 CSA B137.5

## Listings

cNSF®us-pw-G – PEX 5106 ICC-ES PMG 1087 IAPMO UPC PPI TR-4 CSA B137.5 Potable NSF-fs (½" & ¾") cQAlus P371 UL/ULC ANSI/CAN/UL 1821 (excl. ½")





Standards ASTM F876/877, CSA B137.5, and NSF-14 cover the manufacture and testing of all PEX tubing. Multiple parameters are covered within these standards such as workmanship, dimensions and tolerances, density, pressure testing, and the degree of cross-linking. HeatLink® tubing meet and exceed the specified requirements in all cases.

The following quality control tests are completed during the production process.

Tubing Property	Frequency		
Workmanship	Continuously		
Dimensions/Tolerances	Every 30 min		
Density	Daily		
Sustained Pressure	Every 3 months		
Burst Pressure	Every 5 hours		
Environmental Stress Cracking	Yearly		
Degree of Crosslinking	Every 3 days		
Stabilizer Functionality	Yearly		

#### Workmanship:

The surface of the tubing has to be even and smooth. Sharp grooves, sinkholes or other surface disorders must not appear.

#### **Dimensions/Tolerances:**

Diameter and wall thickness is measured according to ASTM dimensions and tolerances for SDR9 PEX tubing. If any deviations beyond the permissible tolerance appear, the tubing is destroyed.

## **Sustained Pressure:**

Tubing samples are tested under pressure for 1000 hours at 200°F (93°C) @ 165psi and 180°F (82.2°C) @ 195psi. HeatLink® tubing meet or exceed the requirements.

## **Burst Pressure:**

Tubing samples are burst within 60 to 70s at 73°F (23°C) to determine the burst pressure. ASTM F876 also specifies burst pressures at additional temperatures. HeatLink® tubing exceeds the requirements at 73°F.

Size	73F	180F	200F
3/8"	620psi	275psi	235psi
1/2"	480psi	215psi	185psi
%" and Larger	475psi	210psi	180psi

## **Environmental Stress Cracking**

One of the most important factors governing the useful life of polyethylene tubing is the formation of stress cracks. Stress cracks can be defined as brittle fractures of the material that normally leads to a creep rupture. Creep rupture is a deformation of the tubing under stress and temperature. Even though the stress level might be below that which would normally cause cracking, stress cracks can appear. The danger increases when the tubing is under additional stress, such as bending during installation.

Stress cracks are greatly reduced in HeatLink's tubing, because of the cross-linked molecular structure, and the unique three-dimensional network. Similarly, accidental scratches that happen on a job site do not affect HeatLink's tubing. Intensive burst pressure tests are performed on tubing after being subjected to scratches as deep as 20% of the tubing wall, show that no damage is caused due to the scratch.



# **Degree of Crosslinking:**

The outstanding properties of HeatLink® tubing is a result of crosslinking the polyethylene. The quality of the tubing depends on the degree of crosslinking. "Degree of crosslinking" represents the percentage of crosslinked material compared to the total mass. This value should not be below 70% for peroxide crosslinked tubing (PEX-a), 65% for vinyl-silane crosslinked tubing (PEX-b), and 65% for electron beam crosslinked tubing (PEX-c). At a lower percentage the characteristics of crosslinked material are not dominant. To test this percentage, a section cut from that sample that represents the entire wall thickness and it is boiled in a solvent, extracting the non-crosslinked portion of the material. Tubing that does not meet the set requirements is destroyed.

#### **Stabilizer Functionality:**

Ensures that the antioxidant package used to protect the tubing is functioning properly. Samples are tested for 3000h at 120°C (248°F) and 101psi (0.70MPa).

#### **PEX Pressure/Temperature Ratings**

All PEX products are evaluated with ASTM D2837 as an indication of the long-term performance of the product. Through evaluation of the test data at The Plastics Pipe Institute (PPI), PEX products are assigned ratings that govern the maximum operating pressure and temperatures.

Temperature	Pressure	HDB	HDS
73F (23C) <sup>1</sup>	160psi (1100kPa) ¹	1250psi	630psi
180F (82C) 1	100psi (690kPa) <sup>1</sup>	800psi	400psi
200F (93C) <sup>2</sup>	80psi (550kPa) <sup>2</sup>	630psi	315psi

<sup>&</sup>lt;sup>1</sup> PureLink Plus, and HeatLink PEX Tubing have been listed to these pressure/temperature ratings as per PPI TR4.

HDB - Hydrostatic Design Basis

HDS - Hydrostatic Design Stress

# **Interpolated & Extrapolated Pressure/Temperature Ratings**

Additional pressure/temperature ratings are determined using a linear interpolation and polynomial trendline extrapolation.

Temperature	Pressure
200.0F (93.3C)	80psi (552kPa)
190.0F (87.8C)	90psi (621kPa)
180.0F (82.2C)	100psi (689kPa)
170.0F (76.7C)	106psi (731kPa)
160.0F (71.1C)	111psi (765kPa)
150.0F (65.6C)	117psi (807kPa)
140.0F (60.0C)	123psi (848kPa)
130.0F (54.4C)	128psi (883kPa)
120.0F (48.9C)	134psi (924kPa)
110.0F (43.3C)	139psi (958kPa)
100.0F (37.8C)	145psi (1000kPa)
90.0F (32.2C)	151psi (1041kPa)
80.0F (26.7C)	156psi (1076kPa)
73.4F (23.0C)	160psi (1103kPa)
60.0F (15.6C)	162psi (1117kPa)
50.0F (10.0C)	162psi (1117kPa)
40.0F (4.4C)	162psi (1117kPa)

The use of PEX tubing in a potable hot-water plumbing system with an operating temperature above 140°F (60°C) or system pressure above 80 psig (550 kPaG) or highly aggressive water quality or any combination thereof can significantly reduce the service life of the tubing. This statement does not apply to closed loop hydronic systems.

HeatLink requires following the guidelines described in Plastics Pipe Institute <u>TN-53</u>, Guide to Chlorine Resistance Ratings of PEX Pipes and Tubing for Potable Water Applications and HeatLink <u>INFO 37</u>, Domestic Hot Water Recirculation Systems.

#### **Excessive Temperature / Pressure Capacity**

In accordance with ASTM F876 Standard Specification for Crosslinked Polyethylene (PEX) Tubing, in the event of a water heating system malfunction, PEX tubing will accommodate short-term conditions of 210°F (99°C) at 150 psi (1034 kPa) of up to 48 hours until repairs can be made. Temperature and pressure relief valves must be installed as per local building codes.

<sup>&</sup>lt;sup>2</sup> PureLink Plus PEX Tubing has been listed to these pressure/temperature ratings as per PPI TR4.





# **Dimensions & Tolerances:**

SDR9 Tubing Dimension Chart (ASTM F876)						
Nominal Size	Outer Diameter		Wall Th	ickness	Average Inn	er Diameter
Nominal Size	mm±tolerance	in±tolerance	mm+tolerance	in+tolerance	mm	in
3/8"	12.70±0.08	0.500±0.003	1.78+0.25	0.070+0.010	8.89	0.350
1/2"	15.88±0.10	0.625±0.004	1.78+0.25	0.070+0.010	12.07	0.475
5/8"	19.05±0.10	0.750±0.004	2.12+0.25	0.083+0.010	14.56	0.574
3/4"	22.22±0.10	0.875±0.004	2.47+0.25	0.097+0.010	17.03	0.671
1"	28.58±0.12	1.125±0.005	3.18+0.33	0.125+0.013	21.89	0.862
1-1/4"	34.92±0.12	1.375±0.005	3.88+0.38	0.153+0.015	26.78	1.054
1-1/2"	41.28±0.16	1.625±0.006	4.59+0.48	0.181+0.019	31.62	1.244
2"	53.98±0.16	2.125±0.006	6.00+0.61	0.236+0.024	41.37	1.629

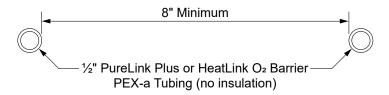


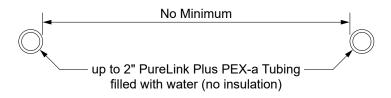
# Flame Spread Index (FSI) & Smoke Developed Index (SDI)

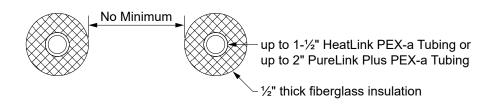
## CAN/ULC S102.2 (Canada)

In compliance with the National Building Code of Canada, HeatLink PEX tubing was tested at QAI and UL in accordance with CAN/ULC S102.2 Standard for Surface Burning Characteristics of Flooring, Floor Covering and Miscellaneous Materials and Assemblies.

	Limitations			Flame Spread	Smoke Developed
Product	Size	Pipe Insulation Requirements*	Spacing Requirements	Index (FSI)	Index (SDI)
HeatLink® O <sub>2</sub> Barrier	1/2"	None	Minimum 8"	< 25	< 50
HeatLink® O <sub>2</sub> Barrier	up to 1-1/2"	rated 1/2" (13 mm)	None	≤ 25	≤ 50
PureLink® Plus	1/2"	None	Minimum 8"	< 25	< 50
PureLink® Plus	up to 2"	rated ½" (13 mm)	None	≤ 25	≤ 50
PureLink® Plus	up to 2" water filled pipe	optional	None	≤ 25	≤ 50







<sup>\*</sup> insulation must be listed to ASTM C547 by an approved agency

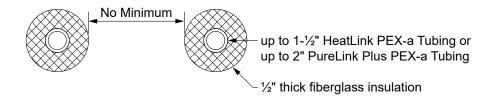


Flame Spread Index (FSI) & Smoke Developed Index (SDI) continued

# **ASTM E84 (United States)**

HeatLink PEX tubing was tested by QAI in accordance with ASTM E84 Standard Test Method for Surface Burning Characteristics of Building Materials.

	Limitations			Flame Spread	Smoke Developed
Product	Size Pipe Insulation Requirements* F		Spacing Requirements	Index (FSI)	Index (SDI)
HeatLink® O <sub>2</sub> Barrier	up to 1-½"	rated 1/2" (13 mm)	None	≤ 25	≤ 50
PureLink® Plus	up to 2"	rated 1/2" (13 mm)	None	≤ 25	≤ 50



<sup>\*</sup> insulation must be listed to ASTM C547 by an approved agency





		Technical Spe	ecifications				
Mechanical Properties		PEX-a (Engel/Peroxide)	PEX-b (Silane)	PEX-c (Irradiation)	Unit	Standard Tested	
Density		930-938	947-950	942-945	kg/m³	ASTM F876	
Flexibility for Installation		Highest	Lowest	Middle			
Tensile strength	@ 20°C	20-26	~20	23-26	N/mm²	DIN 53455	
	@ 100°C	9-13	~12	9-13	N/mm²		
Modulus of elasticity E	@ 20°C	1150	~750	600-900	N/mm²	DIN 53457	
	@ 80°C	560	N/A***	400	N/mm²		
Elongation on failure	@ 20°C	300-450	>300	500-700	%	DIN 53455	
	@ 100°C	500-700	N/A***	750-900	%		
Impact strength	@ 20°C	No failure	No failure	No failure	kJ/m²	DIN 53453	
	@ -140°C	No failure	No failure	No failure*	kJ/m²		
Moisture absorption	@ 22°C	0.01	0.01	0.01	mg/4d	DIN 53472	
Coefficient of friction on steel		0.08-0.1	0.08-0.1	0.08-0.1	-		
Surface energy		34 × 10 <sup>-3</sup>	34 × 10 <sup>-3</sup>	34 × 10 <sup>-3</sup>	N/m		
Oxygen permeability	@ 20°C	0.7 × 10 <sup>-3</sup>	0.7 × 10 <sup>-3</sup>	0.7 × 10 <sup>-3</sup>	g/m³ day	DIN 4726	
	@ 55°C	2.6 × 10 <sup>-3</sup>	2.6 × 10 <sup>-3</sup>	2.6 × 10 <sup>-3</sup>	g/m³ day		
ESCR (environmental cracking)		No failure	No failure	No failure		ASTM F876	
Degree of Crosslinking		70 to 89	65 to 89	65 to 89	%	ASTM F876	
Molecular Crosslink Bond Strength		Carbon-Carbon - 144	Carbon-Silane - 104	Carbon-Carbon - 144	kcal/mol		
Thermal Properties		PEX-a (Engel/Peroxide)	PEX-b (Silane)	PEX-c (Irradiation)	Unit		
Service temperature range		-100, +120	-100, +120	-100, +120	°C		
Coefficient of linear expansion**	@ 20°C	1.4 × 10 <sup>-4</sup>	1.4 × 10 <sup>-4</sup>	1.4 × 10 <sup>-4</sup>	m/m°C		
	@ 100°C	2.05 × 10 <sup>-4</sup>	2.05 × 10 <sup>-4</sup>	2.05 × 10 <sup>-4</sup>	m/m°C		
Softening temperature		+133	+133	+133	°C		
Specific heat		2.3	2.3	2.3	kJ/kg°C		
Coefficient of thermal conductivity		0.38	0.38	0.38	W/m°C		
Electrical Properties		PEX-a (Engel/Peroxide)	PEX-b (Silane)	PEX-c (Irradiation)	Unit		
Specific internal resistance	@ 20°C	10 <sup>15</sup>	10 <sup>15</sup>	10 <sup>15</sup>	-m		
Dielectric constant	@ 20°C	2.3	2.3	2.3	-		
Dielectric loss factor	@ 20°C / 50 Hz	1 × 10 <sup>-3</sup>	1 × 10 <sup>-3</sup>	1 × 10 <sup>-3</sup>	-		
Rupture voltage	@ 20°C	60-90	60-90	60-90	kV/mm		

<sup>\*</sup> tested at - 80°C

<sup>\*\*</sup> Rule of thumb: PEX tubing will expand or contract approximately 1" every 100' of tubing for every 10°F of temperature change.

<sup>\*\*\*</sup> Data not available at time of print



# **PEX Tubing Physical Properties**

PEX Tubing Size	Water Volume Content			ction through the Tubing all*
	US gal/ft	Litres/m	BTU/h/Ft/°F	W/(m °C)
1/2"	0.0092	0.114	5.04	8.70
5/8"	0.0134	0.166	5.17	8.88
3/4"	0.0184	0.228	5.21	8.98
1"	0.0303	0.376	5.19	8.95
1-1/4"	0.0453	0.563	5.20	9.00
1-1/2"	0.0631	0.785	5.17	8.96
2"	0.1083	1.344	5.20	8.97

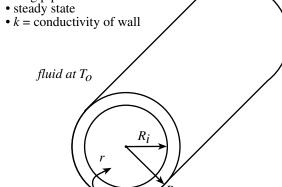
<sup>\*</sup> Coefficient of Thermal Conductivity: 2.635 BTU/h/ft²/(°F/in) 0.38 W/(m °K)

The coefficient of thermal conductivity of PEX has been published virtually in every literature related to PEX tubing. However, the heat conduction rate through the tubing wall can be established for the various tubing sizes as in the following:

Assumptions:

fluid at  $T_i$ 

• long pipe



The heat conduction rate, Q

$$Q = \frac{Ti - To}{R}$$

assuming that fluid in the tubing is at a higher temperature.

$$R = \frac{\ln\left(\frac{Ro}{Ri}\right)}{2\pi kL}$$

R is termed as the thermal resistance of the tubing. L is the tubing length under consideration.

Hence, for unit length of tubing and a  $\Delta T$  (Ti – To) of 1 degree (C or F), the heat conduction rate can be simplified to

$$Q = \frac{2\pi k}{\ln\left(\frac{Ro}{Ri}\right)}$$

where the fluid films on both sides of the tubing wall are considered negligible.

For a CSA  $\frac{1}{2}$ " tubing, Ro = 0.625/2 in, Ri = 0.475/2 in, Q (BTU/Hr/ft/°F)

$$Q = \frac{2 \times \pi \times 0.220}{\ln \left(\frac{0.3125}{0.2375}\right)} = 5.04$$
 where k = 2.635 BTU in/(Hr ft² °F) or 0.220 BTU/(Hr ft °F)

Note: The rate of thermal conduction figures in the table above are for comparison purposes and **not** intended for use in designing floor heating systems. To calculate the heat output of a room other factors must be taken into account (i.e. material encasing the PEX tubing and floor coverings). HeatLink recommends the use of LoopCAD software with the HeatLink Add-on for these calculations.



# **PEX Tubing Pressure Drop Graph**

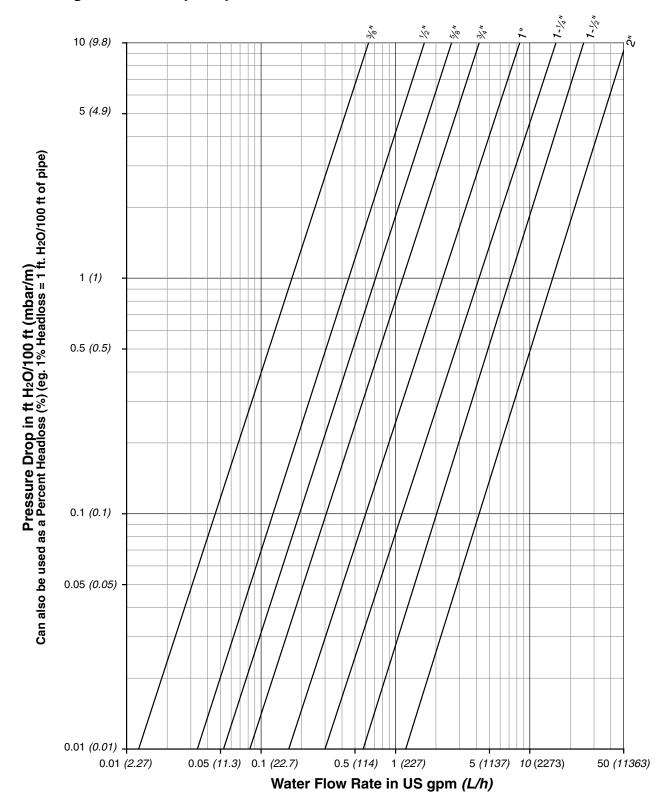


Chart settings at 120°F (49°C)

NOTE: 1 kPa = 10 mbar 100 Pa = 1 mbar 0.0145 psi = 1 mbar 14.5 psi = 1000 mbar 1 ft H<sub>2</sub>O/ft = 97.97 mbar/m 1 mbar/m = 0.010207 ft H<sub>2</sub>O/ft 1 ft H<sub>2</sub>O/ft = 0.4331 psi/ft 1 mbar/m = 0.004421 psi/ft 1 mbar = 0.033456 ft H2O 1 ft H<sub>2</sub>O = 29.89 mbar



# **Oxygen Diffusion**

# What is Oxygen Diffusion?

Oxygen diffusion is the movement of oxygen across the wall of the polyethylene tubing over time. Oxygen can enter a closed loop hydronic heating system in many different ways such as through expansion tanks, air vents, fittings, in the makeup water and through the tubing. Plastic and rubber products are far more porous than metals. While air does not travel through the pores of these materials, an oxygen molecule can. Therefore the oxygen is not visible in the form of bubbles and cannot be eliminated by automatic air vents or air scoops. The operating pressure of the system does not affect the rate of diffusion because the oxygen starved water acts as a negative pressure as far as oxygen is concerned.

When corrosion of ferrous metals occurs in a closed loop heating system the water becomes "oxygen poor." This occurs because the water gives up one of its oxygen molecules to the corrosion reaction. This oxygen poor water then needs to rebuild itself by drawing in oxygen from wherever it can. Oxygen is then pulled in through the porous walls of the polyethylene tubing and once replenished, the corrosion process continues.

The effects of oxygen diffusion in radiant heating systems can be limited to established standards by utilizing tubing with an oxygen barrier. HeatLink PEX tubing is available with an oxygen barrier that allows low gas permeability.

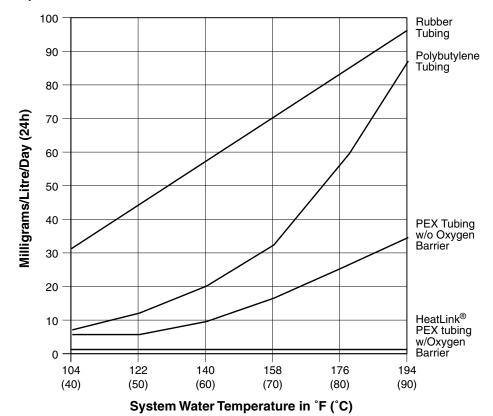
#### What are the answers?

The German industry standards (DIN 4726) have determined that an oxygen diffusion rate of 0.1 milligrams per litre per day @ 40°C in plastic tubing is considered a safe level to prevent corrosion in heating system components. Oxygen diffusion is directly related to the system water temperature (see graph below), the higher the water temperature, the higher the rate of diffusion, which is measured in milligrams per liter per day (24 hours).

For example, the diffusion of 10 milligrams of oxygen per liter (0.001 ounces per gallon) per day through the tubing wall is equivalent to completely draining the heating system and refilling it with fresh water every day during the heating season. This then becomes an open loop system. The HeatLink PEX tubing with oxygen barrier have diffusion rates far below the strict requirements of the German standards and are

recognized as an oxygen tight tubing carrying the DIN label.

In installations where the system will be installed in a "WET" pour, systems operate practically "noiseless." However in "DRY" installations or systems using deflector plates the oxygen barrier can create sound from tubing movements if there are large temperature changes in the supply water temperature. These can be reduced by installation of indoor/outdoor controls. The preferred recommendation for a "Dry" system is to use HeatLink standard PEX tubing. Even so the "DRY" system can not be as quiet as in a "WET" pour. Careful consideration should always be made in suitability of "WET" and "DRY" systems.







# **Oxygen Diffusion**

#### Are there other solutions?

Heat exchangers may be used to isolate the tubing of the radiant system from the ferrous metals in the total system. The heat exchange method eliminates the need for a tube with an oxygen barrier. This alternative adds cost to the installation and therefore may not be economical. Also there are performance restrictions to heat exchangers that may not be acceptable in some installations.

Corrosion inhibitors such as molybdate or nitrite inhibitors must be added to the system to provide total corrosion control. Oxygen is not the only cause of corrosion. Corrosion can also occur in many other forms such as galvanic, caustic or acidic corrosion to name a few. The molybdate and nitrite operate by forming a thin film layer on all metallic parts in the water. This thin film is what provides the protection. See INFO 29 for dosages and details on the corrosion inhibitors. Even with the corrosion inhibitors in the system, oxygen continues to build in concentration but this should not be a concern as long as the inhibitor concentration is

maintained. Homeowners and commercial operators must annually test these levels to ensure that accelerated corrosion does not occur if inhibitors become diluted. Glycol freeze protection fluids can be purchased with inhibitors. These products are important in extreme climate applications and also in HeatLink® SnowMelt installations. A side benefit to these fluids is the lubrication/coating they provide to mixing valves and pumps.

In the above cases HeatLink® PEX may be used. This standard tubing has the same temperature and pressure performance capabilities as all of the HeatLink® PEX tubing that come with oxygen barrier.

For final tubing selections please see your HeatLink® representative who can give expert advice for your application.

# **Kinked Tubing Repair**

One of the most important features of PureLink® crosslinked tubing is its ability to "memorize" its structure and shape. As such, a kinked area can be heated with an electric hot air gun to approximately 260°F (125°C). Please note that open flame can not be used. Heat should be applied evenly until tubing appears clear around its entire circumference. Let the tubing cool undisturbed at room temperature. The repair is now complete.

Note: Colored tubing will not turn clear. O2 barrier will curl back from heated area.

# **Spray foam insulation**

Spray foam insulation can be used in place of batt insulation in DryBelow™ applications. Direct contact of the PEX tubing and spray foam is not allowed. Use a foil barrier to separate them and to keep the foam from getting between the tubing, heat transfer plates, and the subfloor.



- A. Resistant. Can be used within the working pressures.
- B. Conditionally resistant. Restrictions must be made with regards to pressures.
- C. Conditionally resistant. Can be used up to 60% of the working pressures.
- D. Conditionally resistant. Can be used up to 20% of the working pressures.
- U. Not recommended.

		<b>.</b>		
Compound	40°C	Resistance 60°C	80°C	100°C
Accumulator Acid	Α	Α	A	
Acetaldehyde 40%	Α	Α		В
Acetaldehyde 100%	U			
Acetamide	Α	Α	Α	
Acetic Anhydride	Α			
Acetone	Α	Α	Α	
Acetophenone			В	
Acetyl Bromide	U			
Acetyl Chloride			В	
Acetylene	Α	Α	Α	
Acetylene Dichloride	see Dichlo	roethylene		
Acid Mixture H <sub>2</sub> SO <sub>4</sub> -HNO <sub>3</sub> -H <sub>2</sub> O	U	U		
Acid Mixture H <sub>2</sub> SO <sub>4</sub> -H <sub>3</sub> PO <sub>4</sub> -H <sub>2</sub>		В		
Acroline Dispersion	Α			
Acroline Solution	В			
Acryl Nitrite	Α	Α	Α	
Adipic Acid	Α	Α	Α	
Alcohol	see Ethyla			
Aliphatic Esters	A	A	A	
Allyl Alcohol 7%	A	Α	Α	U
Allyl Alcohol 95%	Α			
Allyl Aldehyde	see Acrolir			
Alum	Α	Α	Α	В
Aluminum Acetate	Α	Α	Α	
Aluminum Chloride	Α	Α	A	
Aluminum Fluoride	Α	Α	Α	
Aluminum Hydroxide	A	A	A	
Aluminum Nitrate Solution	Α	Α	Α	
Aluminum Phosphate	A	A	A	
Aluminum Potassium Phosphate	A	A	A	
Aluminum Potassium Sulphate	A	A	A	Α
Aluminum Sodium Sulphate Solution	A	A	A	
Aluminum Sulphate	A	A	A	
Ammonia Aqueous	A	A	A	
Ammonia Gas	A	A	Α	
Ammonium Acetate	A	Α	Α	
Ammonium Aluminum Sulphate	A	Α	Α	
Ammonium Carbonate  Ammonium Chloride	A	A	A	
Ammonium Chloride Ammonium Fluoride 20%	A	A	A	
Ammonium Hydrogen Carbonate	A	A	A	
· ·	A	A	A	
Ammonium Hydrogen Sulphide Ammonium Hydroxide	A	A	A	
Ammonium Metaphosphate	A	A	A	
Ammonium Molybdate	A	А	А	
Ammonium Nitrate	A	Α	Α	
Ammonium Persulphate	A	A	A	
Ammonium Phosphate	A	A	A	
Ammonium Priospriate Ammonium Sulfide	A	A	A	
Ammonium Sulphate	A	A	A	
Ammonium Sulphocyanide	A	, ,		
Ammonium Thiocyanate	A	Α	Α	
Amyl Acetate	A	^	^	
Amyl Alcohol	A	Α	Α	
Amyl Chloride	U	Α	^	
Amyl Methyl Carbinol	В			
Amyl Naphthaline	В			
Aniline Hydrochloride	U			
Aniline Suphate	U			
Aniline (colored)	see Aniline	•		
Aniline (pure)	A	A		
	В			
Aniline (water soluble)	B A	Α	Α	
	B A B	A B	A B	

	Oham	inal Basistana	-	
Compound	Chem 40°C	ical Resistand 60°C	e 80°C	100°C
Anisole		clohexanone	00 C	100 C
Anthroquinone Sulphonic Acid	A	CIONEXAMONE		
Antifreeze Solution	A	Α	Α	
Antimony Pentachloride	A	A	A	
Antimony Trichloride	Α	A	A	
Aqua Regia	U			
Aromatic Acids	A	Α	Α	
Arsenic Acid 80%	A	A	A	
Arsenic Salts	Α		,,	
Arsenic Trichloride	U			
Ascorbic Acid	A			
ASTM Oil no. 1	A	Α	Α	
ASTM Oil no. 2	Α	A	A	
ASTM Oil no. 3	A	A	A	
Atropine Sulphate	Α		,,	
Barium Carbonate	A	A	A	
Barium Chloride	A	A	A	
Barium Hydrosulphide (Bone oil)	,,	,,	В	
Barium Hydroxide	Α	Α	A	
Barium Sulphate	A	A	A	
Barium Suphide	A	A	A	
Beer Colors	A	A	A	
Beer (trading quality)	A	^	Α.	
Benzaldehyde 0.1%	A		С	
	Α		U	
Benzaldehyde 100%  Benzaldehyde Oxime 2%	A			
Benzaldoxime		nzaldahuda O	vimo	
		enzaldehyde O	xime	
Benzene (Benzole)	C	enzoic Acid		
Benzene Carbonic Acid				
Benzene Dicarbonic Acid	see Pr	ntalic Acid		
Benzene Sulphonic Acid			В	
Benzoic Acid	D.		В	
Benzole Carbon Acid		enzoic Acid		
Benzole Dicarbon Acid		ntalic Acid		
Benzole Sulphonic Acid	U			
Benzyl Alechal	В		Р	
Benzyl Alcohol		В	В	
Benzyl Benzoate	Α	ь		
Benzyl Chloride				
Bismuth Carbonate	Α	ΑΑ	A	
Bisulfite		odium Bisulfite	.,	
Bloodstream Salt (red)		otassium Ferric	•	
Bloodstream Salt (yellow)		otassium Ferric	-	
Bonewax	Α		U	
Borax		odium Tetrabor		
Boric Acid	Α	Α	Α	
Boric Copper Sulphate	Α			
Boric Trifluoride	Α			
Brom Oil	Α		В	
Bromethane	U			
Bromine Vapours (low concentration)	U			
Bromine Water	U			
Bromine (liquid)	U			
Butadiene 50%	Α	Α	Α	
Butadiene 100%		В		
Butane (gas)	U			
Butanediol (up to 10%)	Α	Α	Α	
Butanediol (up to 100%)	В			
Butanetriol	Α	Α		
Butanol 100%	Α	Α	Α	
Butanone	Α			
Butene	U			
Butter	Α		В	
Butter Acid	С			
Butter Acid in Water (concentrated)	С			



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	Chemica	l Resistance	9	
Compound	40°C	60°C	80°C	100°C
Butter Acid in Water (solution 20%)	С			
Butyl Acetate			В	
Butyl Acrylate			В	
Butyl Alcohol	see Butar	nol	_	
Butyl Aldehyde	A		В	
Butyl Callulage Solution	U		В	
Butyl Cellulose Solution Butyl Phenol	U			
Butyl Stearate	A	Α	Α	
Butylene	see Buter			
Butylene Glycol	see Butar			
Calcium Acetate	Α	Α	Α	
Calcium Bisulphide	Α			
Calcium Bisulphite	Α	Α	Α	
Calcium Carbonate (soda)	Α	Α	Α	
Calcium Chlorate	Α	Α	Α	
Calcium Chloride	Α	Α	Α	В
Calcium Hydrosulphite (containing SO4)	Α	Α		
Calcium Hydroxide	Α	Α	Α	
Calcium Hypochlorite	Α	Α	Α	
Calcium Nitrate	Α	Α	Α	
Calcium Oxide	A			
Calcium Phosphate	A			
Calcium Sulphate	Α	Α	A	
Calcium Sulphide			В	
Calcium Water	Α		0	
Camphor	Α	Α	C A	
Cane Sugar Juice	A	A	A	
Carbamide 33% Carbolic Acid (phenol)	A	A	A	
Carbon Dioxide	A	A A	A	
Carbon Dioxide (damp)	A	A	A	U
Carbon Dioxide (dry)	A	A	A	A
Carbon Disulphide	С			
Carbon Disulphide	D			
Carbon Monoxide (lamp gas)	Α	Α	Α	
Carbon Tetrachloride	С		U	
Carnbevox	Α			
Caustic Soda	see Sodiu	ım Hydroxide	9	
Cellulose Dissolver	see Ethyle	ene Glycol		
Cetyl Alcohol	Α		В	
Chalk	Α	Α	Α	
Cheese Enzyme	Α	Α	Α	
Chloral Hydrate	Α	Α	Α	
Chloramine	Α	Α	A	
Chlorida Asid		oluene Sulph	no Chlor	
Chloride Acid		ochloric Acid	^	
Chlorine Water 2% Chlorine Water Saturated	A A	Α	A B	
Chlorine (damp gas)	В		n R	
Chlorine (damp gas) Chlorine (dry gas)	В		U	
Chlorine (dry gas) Chlorine (liquid)	U		0	
Chloracetic Acid 85%	В			
Chloracetic Acid 98%	-			U
Chloracetic Acid 100%		В		
Chlorocalcium (in H2O)	Α	A	Α	
Chloroethane	see Ethyl	Chloride		
Chloroethanol	Α	Α	Α	
Chloroethyl Alcohol	see Chlor	oethanol		
Chloroform	С			
Chloromethane	see Methy	yl Chloride		
Chloropropane	see Glyce	erine Chlorhy	drin	
Chlorosulfonic Acid	U			
Chrome Alum	Α	Α	Α	
Chrome Mercury	В			

		Resistance		
Compound	40°C	60°C	80°C	100°C
Chromic Acid 50%	A	Α	A	
Chromic Acid 80%	A		В	
Chromic Acid Anhydride	see Chromium Trioxide see Chromium Trioxide			
Chromium Oxide Chromium Salts		nium Trioxide	!	
Chromium Saits Chromium Trioxide 20%	A	Α	A	
Chromium Trioxide 50%	A	А	В	
Chromium Trioxed 80%	Α		Ь	
Cider	A			
Cis-Oxime		dehyde Oxim	٩	
Coal Tar	U	acriyac Oxiiii		
Cobalt Chloride	A	Α	Α	
Cocoa Fat	Α	Α	Α	
Cocoa Fat Alcohol	Α	Α	Α	
Coconut Oil	Α	Α	Α	
Coffee	Α			
Colanut (concentrated)	Α	Α	Α	
Cooking Salt	see Sodiu	m Chloride		
Copper Acetate			В	
Copper Chloride (cupric)	Α	Α	Α	
Copper Chloride (cuprous)	Α	Α	Α	
Copper Cyanide	Α	Α	Α	
Copper Fluoride	Α	Α	Α	
Copper Nitrate	Α	Α	Α	
Copper Sulphate	Α	Α	Α	
Corn Oil	Α	Α	Α	
Cotton Seed Oil	Α	Α	Α	
Creosote	U			
Cresol	U			
Cyanides	Α	A	A	
Cyclohexane	C			
Cyclohexanol	A C		В	
Cyclohexanone DDT	A		Ь	
Decahydro Naphthalene	C		В	
Denatured Spirit	see Methy	I Alcohol		
Dextrine	A	A	Α	
Dextrose	A	A	A	Α
Diacetone	Α	Α	Α	
Diacetone Alcohol	Α			
Diammonium Salts	Α	A	Α	
Dibenzyl Ether			В	
Dibutyl Ether	Α	Α	Α	Α
Dibutyl Phthate	Α		В	
Dibutyl Sebacate	Α		В	
Dichloracetic Acid	Α	Α	Α	
Dichloracetic Acid Methylester	Α	Α	Α	
Dichlorobenzene	В			
Dichloroethane	see Ethyl (	Chloride		
Dichloroethylene	U			
Dichloromethane	see Methy	l Chloride		
Dichlorohexamine			В	
Diesel Oil			С	
Diethylene Glycol Monobutylene	Α		_	
Diethyl Benzene	= 0		В	
Diethyl Ether	see Ethyl I	±tner		
Diethyl Phthalate	Α		В	
Diethylana Dievide	and District	20	В	
Diethylene Dioxide	see Dioxa			
Diethylene Glycol	A	A	A	
Diglycol Acid	A	A	A	
Dihexyl Phthalate Diisobutylene	^	^	В	
Diisopropyl Ketone	A	Α	A	
Dimethyl Amine	A	^	^	
Dimodiff Amilio	^			



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	Chemical	Resistance		
Compound	40°C	60°C	80°C	100°C
Dimethyl Aniline			В	
Dimethyl Benzole	see Xylol			
Dimethyl Formamide	Α	Α	Α	
Dimethyl Ketone	see Aceto	ne		
Dimethyl Phthalate	Α	Α	Α	
Dioctyl Phthalate			В	
Dioctyl Sebacate			В	
Dioxalane			В	
Dioxane			С	
Dioxyethyl Ether	see Diethy	lene Glycol		,
Diphenyl			В	
Diphenyl Amine	A			
Diphenyl Oxide	U			
Disodium Phosphate	A			
Edible Oil	A	A	A	
Electrolyte 10% Elementine (normal concentration)	A	A	A	Α
	A	A	A	А
Emulsions (photographic)	А	А	В	
Epichlorohydrin Epoxy Ethane	see Ethyle	no Ovido	Ь	
Esteric Oils	B Ethyle	B Dxide	В	
Ethanal	see Acetal		Ь	
Ethandiol	see Ethyle	•		
Ethane Diamine		ne Diamine		
Ethanol	see Ethyl			
Ethanolamine	See Luiyi7	-iconoi	В	
Ether	see Ethyl	Ethor	Ь	
Ethoxyethane	see Ethyl			
Ethyl Acetate	SCC Lulyi	Luici	С	
Ethyl Alcohol	Α		A	В
Debatured with 2% Toluol	A			
plus Acetic Acid (quality use)	A			
Ethyl Benzene	В			
Ethyl Benzoate	В			
Ethyl Carbitol			В	
Ethyl Cellulose			В	
Ethyl Chloride	Α			
Ethyl Ether	С		U	
Ethyl Formate			В	
Ethyl Glycol			В	
Ethyl Methyl Ketone	see Butan	one		
Ethyl Oxalate	Α	Α	Α	
Ethyl Pentachloro Benzene	U			
Ethyl Salicylate	В			
Ethyl Silicsate	Α	Α	Α	
Ethyl Valeriate	Α			
Ethylamine	Α	Α	Α	
Ethylene Chlorohydrin	U (see Ch	loroethanol)		
Ethylene Chloride	В			
Ethylene Diamine	Α			
Ethylene Dichloride	В			
Ethylene Glycol 100%	Α	Α	Α	В
trading quality		Α	Α	U
Ethylene Glycol Monoethyl Ether	A			
Ethylene Oxide (liquid)	U			
Ethylene Trichloride	D			
"Eugenol"	В			
Fatty Acid	Α	de La cal II	В	
Ferric Chloride	see Iron C			-
Fertilizer Salts	Α	Α	A	В
Fish Oil	A	Α	Α	
Fluorbenzene	U			
Fluorboric Acid	A	A	A	
Fluorides Fluorine (solution)	A	Α	Α	
CONTRACTOR (SOURION)	U			

Compound		I Resistance		40000
Compound	40°C	60°C	80°C	100°C
Formaldehyde 40% Formaldehyde (diluted)	A A	A	Α	
Formamide	A	A	A	
Formic Acid	A	A	A	
Freon 12	Α	^	В	
Freon 13	Α	Α	A	
Freon 21	U	,,	,,	
Freon 22	A	Α	Α	
Freon 114	A	A	A	
Fruit Juice	Α	Α	Α	U
Fruit Mass	A	A	A	
Fruit Sugar	Α			
Fuming Sulphuric Acid	see Oleui	m		
Furan	D			
Furfural	В		U	
Furfural Alcohol	Α			
Gas (natural)	Α			
Gases Containing				
Carbon Dioxide, Carbonic				
Acid (all concentrations)	Α	Α	Α	Α
Chlorine (all concentrations)	Α	Α	Α	В
Fluorine Traces	Α	Α	Α	U
Nitrous Oxide Traces	Α	Α	Α	U
Oleum (low concentrations)	U			
Sulphur Dioxide 50%	Α	Α		
Sulphur Dioxide (low conc.)	Α	Α	Α	В
Sulphuric Acid (all conc.)	Α	Α	Α	
Gasoline (pure; 100 octane)	С	В		
Gasoline/Benzene mixture (80/20)	С		В	
Gelatine	Α	Α	Α	
Gin	В			
Glucose	see Dexti	ose		
Glycerine Chlorhydrin	Α	Α	Α	
Glycerine, Glycerol	Α	Α	Α	
Glycine	see Glyco	ol		
Glycol 10%	Α	Α		
Glycol Dichloride	see Ethyl	ene Chloride		
Glycol Ester	Α	Α	Α	
Glycolic Acid 37%	Α	Α	Α	
Heating Oil, Barrel Oil	Α			
Heavy Emulsion	see Bariu	m Carbonate		
Heavy Oil			В	
Hydrochloric Acid <30%	Α	Α	Α	U
Hydrochloric Acid >30%	Α		В	U
Hydrochloric Acid 10%	Α	Α	Α	U
Hydrochloric Acid 20%	Α	Α	В	U
Hydrocyanic Acid	see Hydro	ogen Cyanide	•	
Hydrogen	Α	Α	Α	
Hydrogen Bromide	Α	Α	Α	
Hydrogen Chloride Gas	Α	Α	Α	
Hydrogen Cyanide	Α	Α	Α	
Hydrogen Fluoride 40%	Α	Α		
Hydrogen Fluoride 70 %	Α			
Hydrogen Peroxide 30%	Α	Α	Α	
Hydrogen Peroxide 90%	Α	Α	Α	
Hydrogen Sulphide (dry)	Α	Α	Α	
Ink	Α	Α	Α	
lodine Ink	Α			
lodine (alcoholic solution)	В			
Iodine-Potassium Iodide 3%				
Iodine	Α		U	
Iron (II) Chloride	Α	Α	Α	
1 1				
Iron (II) Sulphate	Α	Α	Α	
1 1	A A A	A A A	A A A	Α



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	Chemica	l Resistanc	е	
Compound	40°C	60°C	80°C	100°C
Iron (III) Sulphate	Α	Α	Α	
Ketones	В			
Lacquer	U			
Lactic Acid 90%	Α	Α	Α	
Lanolin	Α	Α	Α	
Latex	A			
Lauryl Alcohol	В			
Lavender Oil			В	Б
Lead Acetate Lead Arsenate	A A	Α	Α	В
Lead Nitrate	A	A	A	
Lead Sulphamate	A	A	A	
Lemon Oil	В	7.	U	
Linseed Oil	A		В	
Liquor (trading quality)	Α			
Machine Oil	Α	В		
Magnesium Carbonate	Α	Α	Α	
Magnesium Chloride	Α	Α	Α	
Magnesium Hydroxide	Α	Α	Α	
Magnesium Iodine	Α			
Magnesium Nitrate	Α	Α	Α	
Magnesium Sulphate	Α	Α	Α	Α
Maleic Acid	Α	Α	Α	Α
Manganese Sulphate	Α	Α	Α	
Marmelade	Α	Α	Α	
Melase spices (industrial concentration)	Α	Α	Α	
Melase (industrial concentration)	Α	Α	Α	Α
Menthanol	see Ment			
Menthol	A	A	A	
Mercury	Α	Α	Α	
Mercury Salts	Α	Α	Α	В
Mesityl Oxide Methane			В	В
Methane Amide	see Form	aldehyde	ь	
Methanol		yl Alcohol		
Methoxy Butanol	A	A	A	
Methyl Acetate	, ,	,,	В	
Methyl Alcohol	Α	Α	Α	
Methyl Amine 32%	Α			
Methyl Bromide	see Brom	ethane		
Methyl Butyl Ketone	A	Α	Α	
Methyl Cellulose Solvent	Α			
Methyl Chloride	D			
Methyl Ethyl Ketone	D			
Methyl Formate			В	
Methyl Glycol	Α	Α	Α	
Methyl Isobutyl Ketone			С	
Methyl Methacrylate			В	
Methyl n-Propyl Ketone	В			
Methyl Oleate	A	. A	A	
Methyl Phenol	see Cres	OI		
Methyl Salicate	В	В		
Methyl Sulphate (acid)	A	B vl Sulphurio	Aoid	U
Methyl Sulphate (acid) Methylene Chloride	See Meth D	yl Sulphuric	Aciu	
Milk	A	A	A	
Mineral Oil	A	, ,	В	
Mineral Water	A	Α	A	
Monochloride Acetic Acid	A	A	A	
Monochloride Acetic Acid Ethylester	A	A	A	
Monochloride Acetic Acid Methylester	A	A	A	
Monochloro Benzene	D			
Monoethanolamine	Α			
Monoethyl Ether	Α	Α	Α	
Monomethyl Aniline	Α	Α	Α	

		Resistance		
Compound	40°C	60°C	80°C	100°C
Motor Oil	^		С	
Naphthalene, Naphthaline Natural Gas	A U		В	
Nicotine	A			
Nicotinic Acid	A			
Nitric Acid <30%	A	A		
Nitric Acid 30-50%	, .	C		
Nitric Acid 40%			U	
Nitric Acid 70%	С			
Nitric Acid 98%			U	
Nitroethane	Α		U	
Nitrogen	Α	Α	Α	
Nitromethane	Α		U	
Nitrotoluol	Α		В	
Nitrous Gases (concentrated)	Α		U	
Nonyl Alcohol	A		_	
Octyl Alcohol	A		В	
Octyl Cresol	В	0	U	
Oil Oil Acid	С	С	С	
Oleum Vapour (SO3)	В		-	
Olive Oil	A	Α	Α	
Ortho-Boric Acid	see Boric A		^	
Oxalic Acid	A	A	Α	
Oxyacetic Acid	see Glycoli		,,	
Oxybenzole	see Pheno			
Oxydiethanole	see Diethyl			
Oxypropionic Acid	see Lactic	Acid		
Oxyrane	see Ethylei	ne Oxide		
I-Oxytoluol	see Benzyl	Alcohol		
m-Oxytoluol	see Cresol			
Ozone	В		U	
Painting Turpentine	see Thinne	r		
Palmatic Acid			В	
Palmolive Oil	Α		_	
Paraffin		_	В	
Paraffin emulsion (trading quality)		В		
Paraffin Oil	Α	Α	Α	
Paratoluene Sulpho Chloro Amide Sodium 1%	A			
Pentanol	see Amyl A	Icohol		
Pentanol Acetate	see Amyl A			
Perchloric Acid 20%	Α	A	Α	
Perchloric Acid 50%	Α		В	
Perchloro Ethylene	U			
Petrol Ether			С	
Petroleum			С	
Phenol (up to 90%)	Α	Α		U
Phenyl Alcohol	see Benzyl	Alcohol		
Phenyl Benzene				
Phenyl Ethane	see Ethyl E			
Phenyl Hydrazine	Α	Α	Α	
Phenyl Hydrazine Hydrochloride				U
Phenyl Methane	see Toluol			
Phenyl Methyl Ether	see Cycloh	exanone		
Phosgene (gas) Phosphates	A A	٨	Α	
	A	A	A	U
Phosphoric Acid 80% Phosphorus Oxychloride	A	В	В	J
Phosphorus Pentaoxide	A	A	A	
Phosphorus Trichloride	A	^	В	
Photographic Solution (fixer)	A	Α	A	
Phthalic Acid 50%	A	A	A	
Picric Acid 1%	A	• •	В	
Pinene			В	



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	Chem	ical Resistance			
Compound	40°C	60°C	80°C	100°C	
Potassium Acetate			В		
Potassium Bichromate 40%	see Po	otassium Dichrom	ate		
Potassium Borate 1%	Α	Α	Α		
Potassium Bromate	Α	Α	Α		
Potassium Bromide	A	Α	A		
Potassium Carbonate	A	A	A		
Potassium Chlorate Potassium Chloride	A A	A	A		
Potassium Chromate	A	A A	A A		
Potassium Chromium Sulphate	A	A	A	В	
Potassium Cupro Cyanide	A	A	A		
Potassium Cyanide	Α	A	Α		
Potassium Dichromate 40%	Α	Α	Α		
Potassium Ferricyanide	Α	Α	Α	В	
Potassium Fluoride	Α	Α	Α		
Potassium Hydrogen Carbonate	Α	Α	Α		
Potassium Hydrogen Sulphate	Α	Α	Α		
Potassium Hydrogen Sulphite Solution	Α	Α	Α		
Potassium Hydroxide 50%				U	
Potassium Hydroxide 60%	Α	Α	Α		
Potassium Hypochlorite Solution	Α		В		
Potassium lodide (cold saturated)	Α	Α	Α		
Potassium Nitrate	Α	A	A		
Potassium Orthophosphate	A	A	A		
Potassium Perborate	Α	A	A		
Potassium Perchlorate 1%  Potassium Perchlorate 10%	A	Α	Α	Α	
	A A	۸	Α		
Potassium Permanganate 18% Potassium Phosphate	A	A A	A		
Potassium Sulphate	A	A	A		
Potassium Sulphate (cold saturated)	A	A	A		
Potassium Sulphide	A	A	Α		
Potassium Sulphite	A	,,	,,		
Potassium Supersulphate	Α	Α	Α	U	
Pseudo Cumol	Α	Α	Α		
Propane Acid	see Pi	ropionic Acid			
Propane Diol	see Pi	ropylene Glycol			
Propane Triol	see G	lycerine			
Propane (gas)	Α				
Propane (liquid)	В				
Propanol	Α	Α	Α		
Propanone		cetone			
Propene	A	A	A		
Propionic Acid	Α	Α	A		
Propyl Alcohol	see D	ronanol	В		
Propyl Alcohol Propylene Dichloriole	U See Pi	ropanol			
Propylene Glycol	A	Α	Α		
Propylene Oxide	A	, ,	,,		
Pyridine	.,		С		
Pyrol			В		
Resorcinol	Α				
Ricine Oil	Α		В		
Rinser Loosener	Α	Α	Α		
Road Tar	U				
Salicylic Acid	Α	Α	Α		
Selenic Acid	Α				
Silicone Fats	Α	Α	Α		
Silicone Oils	Α	Α	Α		
Silver Nitrate <80%	Α	A	Α	В	
Silver Salts (cold saturated)	A	A	A		
Soap	A	A	A		
Soap Loosener	A	A	A		
Soap Solution	A	A adii wa Carbanata	Α		
Soda	see Sodium Carbonate				

Compound	Chemical 40°C	Resistance 60°C	80°C	100°C
Sodium Acetate	Α	Α	Α	100 C
Sodium Aluminate	A	A	A	
Sodium Benzoate	A	A	Α	
Sodium Benzoate (up to 36%)	Α	Α	Α	
Sodium Bicarbonate	Α	Α	Α	
Sodium Bisulphate	A	A	A	
Sodium Bisulphite	Α	Α	Α	Α
Sodium Borate	Α	Α	Α	
Sodium Bromide	Α	Α	Α	
Sodium Carbonate	Α	Α	Α	
Sodium Chlorate	Α	Α	Α	
Sodium Chloride	Α	Α	Α	Α
Sodium Chlorite and Bleach	Α		В	
Sodium Chlorite and Water	Α			
Sodium Cyanide	Α	Α	Α	
Sodium Ferricyanide	Α	Α	Α	
Sodium Ferrocyanide	Α	Α	Α	
Sodium Fluoride	Α	Α	Α	
Sodium Hydrogen Carbonate	Α	Α	Α	
Sodium Hydrogen Phosphate	Α	Α	Α	
Sodium Hydrogen Sulphite Solution	Α	Α	Α	
Sodium Hydrosulphite 10%	Α	Α	Α	
Sodium Hydroxide	Α	Α	Α	
Sodium Hypochlorite	Α	Α	Α	
Sodium Hyposulphate				
Sodium Metaphosphate	Α	Α	Α	
Sodium Nitrate	Α	Α	Α	
Sodium Nitrite	Α	Α	Α	
Sodium Perborate	Α	Α	Α	
Sodium Peroxide	A	Α	A	
Sodium Phosphate	Α	Α	Α	
Sodium Polycrylate (GR 894)	A	A	A	
Sodium Silicate	A	A	A	
Sodium Sulphate	A	A	A	
Sodium Sulphide	A	A	Α	
Sodium Sulphite Sodium Tetraborate	A	A	A A	
	A	A	A	
Sodium Thiosulphate	A	A	A	
Soya Oil	A		В	
Spinning Oil Spinning Bath Oil Containing			ь	
Carbon Disulphide 0.01%	Α	Α		
Spinning Bath Oil Containing	A	^		
Carbon Disulphide 0.07%	Α	Α		
Spot Solvents	A	A	Α	
Starch	A	A	A	
Starch Syrup	A	A	A	Α
Stearic Acid			В	
Sucrose Solution	Α	Α	A	
Sulphur	A	A	A	
Sulphur Dioxide (dry)	Α	A	Α	В
Sulphur Dioxide (in water solution)	Α	Α	Α	
Sulphur Solution	Α			
Sulphur Trioxide	С			
Sulphuric Acid <50%	Α	Α	Α	
Sulphuric Acid 70%	Α		С	
Sulphuric Acid 80-90%		С		
Sulphuric Acid 96%			С	
Sulphuryl Chloride	В			
Superchloric Acid	see Perch	loric Acid		
Synthetic Washing Powder				
(home quality)	Α	Α	Α	
Tannin	see Ascor	bic Acid		
Tar	U			
Tertiary Butyl Alcohol	Α	Α	Α	



- A. Resistant. Can be used within the working pressures.
- B. Conditionally resistant. Restrictions must be made with regards to pressures.
- C. Conditionally resistant. Can be used up to 60% of the working pressures.
- D. Conditionally resistant. Can be used up to 20% of the working pressures.
- U. Not recommended.

Compound	Chemic 40°C	al Resistano 60°C	e 80°C	100°C	
Tetrachloro Ethane	C	00 0	U	100 0	
Tetraethyl Lead	A		, ,		
Tetrahydro Furane	С		С		
Tetrahydro Furfuyl Alcohol	Α				
Tetrahydro Naphthalene	D		D		
Tetraline	see Tetra	ahydro Naph	thalene		
Tetramethylene Oxide		ahydro Furar	ne		
Thinner	D				
Thionyl Chloride Thiophen	U B		В		
Tin Salts	A	Α	A		
Titanium Tetrachloride	U	,,	,,		
Touene	С		С		
Transformer Oil	С		D		
Tri	see Tricl	hloro Ethylen	ie		
Tributro Ethyl Phosphate			В		
Tributyl Phosphate	Α	Α	Α		
Trichloro Acetic Acid	Α		В		
Trichloro Acetic Acid 50%	Α	Α	Α		
Trichloro Benzene	U				
Trichloro Ethane	С		U		
Trichloro Ethylene Trichloro Methane	C see Chlo				
Trivresyl Phosphate	see Chic	DIOIOIIII	В		
Triethanolamine	Α		Ь		
Triethylene Glycol	A				
Trilom (trade quality)	Α	Α	Α		
Trimethyl benzene	see Pse	udocumol			
Trinitro Phenol	see Picr	ic Acid			
Trinitro Toluene	U				
Trioctyl Phosphate	Α		В		
Trisodium Phosphate	Α	Α	Α		
Turbine Oil	•		В		
Turpentine Uric Acid	C A		С		
Uric Compounds	see Carl	hamida			
Urine (normal concentration)	A	A	Α		
Vaseline	, ,		В		
Vaseline Oil	Α		В		
Vegetable Oils	В	В	В		
Vinegar	Α	Α	Α		
Vinegar Acid Anhydride	Α	Α	В	U	
Vinegar Acid Butyl Ester	see Buty	/I Acetate			
Vinegar Acid Ethyl Ester		/I Acetate			
Vinegar Ester		/I Acetate			
Vinegar (trading quality)	A	A	A		
Vinyl Chloride	A see Acry	A ⁄I	Α		
Vinyl Cyanide	Nitrite				
			Α	Α	
Water	A	Α		^	
Waterglass	Α				
Waterglass Whisky	A (see E	thol Alcohol)		-	
Waterglass Whisky Whitener	A (see E see Sod	thol Alcohol)			
Waterglass Whisky Whitener Wine Vinegar	A (see E see Sod see Vine	ithol Alcohol) ium Hypochl	orite		
Waterglass Whisky Whitener	A (see E see Sod	thol Alcohol)	orite A	n	
Waterglass Whisky Whitener Wine Vinegar Wine (red and white)	A (see E see Sod see Vine A	ithol Alcohol) ium Hypochl egar A	orite		
Waterglass Whisky Whitener Wine Vinegar Wine (red and white) Wohlstone Acid	A (see E see Sod see Vine A A	ithol Alcohol) ium Hypochl egar A A	orite A		
Waterglass Whisky Whitener Wine Vinegar Wine (red and white) Wohlstone Acid Wood Glue (type polyvinyl acetate)	A (see E see Sod see Vine A A B	ithol Alcohol) ium Hypochl egar A A	orite A		
Waterglass Whisky Whitener Wine Vinegar Wine (red and white) Wohlstone Acid Wood Glue (type polyvinyl acetate) Wool Fat	A A (see E see Sod see Vine A A B see Lane	ithol Alcohol) ium Hypochl egar A A	A A		
Waterglass Whisky Whitener Wine Vinegar Wine (red and white) Wohlstone Acid Wood Glue (type polyvinyl acetate) Wool Fat Xylol Yeast Zinc (II) Chloride	A A (see E see Sod see Vine A A B see Land C A A	ithol Alcohol) ium Hypochl egar A A Ollin A A	A A U A A	В	
Waterglass Whisky Whitener Wine Vinegar Wine (red and white) Wohlstone Acid Wood Glue (type polyvinyl acetate) Wool Fat Xylol Yeast Zinc (II) Chloride Zinc Carbonate	A A (see E see Sod see Vine A A B see Land C A A A	ithol Alcohol) ium Hypochl egar A A  bolin A A	A A U A A A		
Waterglass Whisky Whitener Wine Vinegar Wine (red and white) Wohlstone Acid Wood Glue (type polyvinyl acetate) Wool Fat Xylol Yeast Zinc (II) Chloride Zinc Carbonate Zinc Hydrate	A A (see E see Sod see Vine A A B see Land C A A A A A A	ethol Alcohol) ium Hypochl egar A A  ollin A A A	A A U A A A A		
Waterglass Whisky Whitener Wine Vinegar Wine (red and white) Wohlstone Acid Wood Glue (type polyvinyl acetate) Wool Fat Xylol Yeast Zinc (II) Chloride Zinc Carbonate Zinc Hydrate Zinc Oxide	A (see E see Sod see Vine A A B see Land A A A A A A A A	ithol Alcohol) ium Hypochl egar A A  bolin A A	A A U A A A		
Waterglass Whisky Whitener Wine Vinegar Wine (red and white) Wohlstone Acid Wood Glue (type polyvinyl acetate) Wool Fat Xylol Yeast Zinc (II) Chloride Zinc Carbonate Zinc Hydrate	A A (see E see Sod see Vine A A B see Land C A A A A A A	ethol Alcohol) ium Hypochl egar A A  ollin A A A	A A U A A A A		